

Chemical Engineering Thesis Defense

Carbon dioxide transfer characteristics of hollow-fiber, composite membranes

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abstract

Carbon dioxide (CO₂) levels in the atmosphere have reached unprecedented levels due to increasing anthropogenic emissions and increasing energy demand. CO₂ capture and utilization can aid in stabilizing atmospheric CO₂ levels and producing carbon-neutral fuels. Utilizing hollow fiber membranes (HFMs) for microalgal cultivation accomplishes that via bubbleless gas-transfer, preventing CO₂ loss to the atmosphere. Various lengths and geometries of HFMs were used to deliver CO₂ to a sodium carbonate solution. A model was developed to calculate CO₂ flux, mass-transfer coefficient (KL), and volumetric mass-transfer coefficient (KLa) based on carbonate equilibrium and the alkalinity of the solution. The model was also applied to a sparging system, whose performance was compared with that of the HFMs. Typically, HFMs are operated in closed-end mode or open-end mode. The former is characterized by a high transfer efficiency, while the latter provides the advantage of a high transfer rate. HFMs were evaluated under both modes of operation and a varying inlet CO₂ concentration to determine the effect of inert gas and water vapor accumulation on transfer rates. It was found that for pure CO₂, a closed-end module operates as well as an open-end module. Closed-end modules perform significantly worse when CO₂-enriched air is supplied. This was shown by the KLa values calculated using the model. Finally, a mass balance was conducted on the membranes in order to develop a model that provided insight into the gas concentration profiles inside the fiber lumen. For dilute CO₂ inlet streams, accumulation of inert gases like nitrogen (N₂), oxygen (O₂), and water vapor (H₂O) significantly affected module performance by reducing the average CO₂ partial pressure in the membrane and diminishing the amount of interfacial mass-transfer area available for CO₂ transfer.

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